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Evidence for Cyclophosphate Formation During Hydrolysis of 3-MethylcycloSal-PCVMP

Andreas Lomp^a; Chris Meier^a; Markus Herderich^b; Peter Wutzler^c
^a Institut für Organische Chemie, Universität Hamburg, Hamburg, Germany ^b Institut für Lebensmittelchemie, Universität Würzburg, Würzburg, Germany ^c Institut für Antivirale Chemotherapie, Klinikum der Friedrich-Schiller-Universität Jena, Erfurt, Germany

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EVIDENCE FOR CYCLOPHOSPHATE FORMATION DURING HYDROLYSIS OF 3-METHYL-CYCLOSAL-PCVMP

Andreas Lompa, Chris Meiera*, Markus Herderichb and Peter Wutzlerc

- a) Institut für Organische Chemie, Universität Hamburg, Martin-Luther-King-Platz 6, D-20146 Hamburg, Germany
 - b) Institut für Lebensmittelchemie, Universität Würzburg, Am Hubland D-97074 Würzburg, Germany
- c) Institut für Antivirale Chemotherapie, Klinikum der Friedrich-Schiller-Universität Jena, Nordhäuser Str. 78, D-99089 Erfurt, Germany

Abstract: A hydrolysis study of 3-methyl-cycloSal-PCVMP 2 is described. Surprisingly, phosphotriester 5 released in this study not the expected PCVMP, but cycloPCVMP.

Here we describe the properties of 3-methyl-cyclosaligenyl penciclovirmono-phosphate 2. Penciclovir 1 and the title compound 2 were prepared according to syntheses that have been described previously 1,2.

Compound 2 has been studied for its lipophilic property (partition coefficient in n-octanol/water (logP value)) as well as for its hydrolytic stability in 25 mM isotonic PBS, pH 7.3. For this hydrolysis experiment we proceeded as described before³. The logP value, the half-life (t_{1/2}) and the rate constant k have already been published².

Surprisingly, 3-methyl-cycloSal-PCVMP 2 showed a pronounced decrease in half-life as compared to other cycloSal-NMP triesters². We assumed that the reason for this marked increase in hydrolytic lability could be either an assisting effect of the second, unprotected primary hydroxy group in the vicinity of the phosphate triester moiety due to a preorientation or an activation of the nucleophile water or a direct nucleophilic attack of this hydroxyl group at the phosphorus center. The consequences are i) the formation of PCVMP in the former case (except the rate acceleration) and ii) the formation of cyclic PCVMP 3 (cPCVMP). In order to identify the hydrolysis product of the chemical hydrolysis of 3-methyl-cycloSal-PCVMP 2 two experiments were conducted: First, a ³¹P-NMR experiment in a DMSO-d₆/water mixture containing small amounts of TRIS-buffer,

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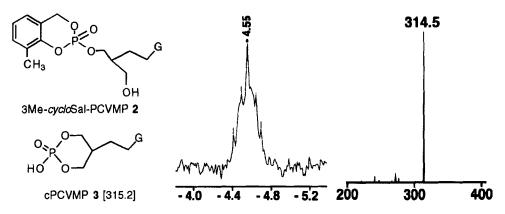


Figure 1: Structures of 2 and 3, H-coupled ³¹P-NMR and ESI-mass spectrum of hydrolysis product 3

pH 8.8 was carried out. After complete hydrolysis, a proton-coupled ³¹P-NMR spectrum was recorded. From the chemical shift and the coupling pattern of the phosphorus signal (a singulet at -4.5 ppm in the decoupled mode) it became apparent that the hydrolysis of 2 obviously did not lead to the expected PCVMP but to a product which showed a quintet in the proton-coupled mode (Figure 1). In contrast, PCVMP should split into a triplet in the proton-coupled mode of the ³¹P-NMR spectrum⁴. Consequently, the multiplet observed strongly indicates the formation of the cPCVMP 3 (Figure 1). A further indication for the formation of cPCVMP 3 was taken from electrospray mass spectrometry. PCVMP has a mass of 333.2 u and should yield molecular ions [M-H]⁻ and [M-2H]⁻ at m/z 332 and m/z 165.5, respectively. However, the product observed in the NMR experiment was characterized by a molecular ion [M-H]⁻ at m/z 314 (Figure 1). Further product ion experiments unambiguously demonstrated the presence of a phosphate moiety (m/z 97 and m/z 79). Hence, both experiments prove the formation of cPCVMP 3 (MW 315.2 g/mol) from the *cyclo*Sal-PCVMP derivatives 2 instead of PCVMP.

Compound 2 has been evaluated for its inhibition of HSV-1, HSV-1/TK⁻ and EBV replication in Vero and P3HR-1 cells. As expected, 2 devoid of any antiviral activity².

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